

A Correlation of
Elevate Science: Physical
©2019



To the
Idaho Content Standards for Science
Physical Sciences
Grades 6-8

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Idaho Content Standards for Physical Sciences, Grades 6-8

Introduction

This document demonstrates how **Elevate Science ©2019** meets the Idaho Content Standards for Science. Correlation page references are to the Student and Teacher’s Editions and cited at the page level.

Savvas is proud to introduce **Elevate Science** Middle Grades – where exploration is the heart of science! Designed to address the rigors of new science standards, students will experience science up close and personal, using real-world, relevant phenomena to solve project-based problems. Our newest program prepares students for the challenges of tomorrow, building strong reasoning skills and critical thinking strategies as they engage in explorations, formulate claims, and gather and analyze data that promote evidence-based arguments. The blended print and digital curriculum covers all Next Generation Science Standards at every grade level.

Elevate Science helps teachers transform learning, promote innovation, and manage their classroom.

Transform science classrooms by immersing students in active, three-dimensional learning.

Elevate Science engages students with real-world tasks, open-ended Quests, uDemonstrate performance-based labs, and in the engineering/design process with uEngineer It! investigations.

- A new 3-D learning model enhances best practices.
- Engineering-focused features infuse STEM learning.
- Phenomena-based activities put students at the heart of a Quest for knowledge.

Innovate learning by focusing on 21st century skills.

Students are encouraged to think, collaborate, and innovate! With **Elevate Science**, students explore STEM careers, experience engineering activities, and discover our scientific and technological world. The content, strategies, and resources of Elevate Science equip the science classroom for scientific inquiry and science and engineering practices.

- Problem-based learning Quests put students on a journey of discovery.
- STEM connections help integrate curriculum.
- Coding and innovation engage students and build 21st century skills.

Manage the classroom with confidence.

Teachers will lead their class in asking questions and engaging in argumentation. Evidence-based assessments provide new options for monitoring student understanding.

- Professional development offers practical point-of-use support.
- Embedded standards in the program allow for easy integration.
- ELL and differentiated instruction strategies help instructors reach every learner.
- Interdisciplinary connections relate science to other subjects.

Designed for today's classroom, preparing students for tomorrow's world. **Elevate Science** promises to:

- Elevate thinking.
- Elevate learning.
- Elevate teaching.

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PS Physical Sciences	
PS1-MS Matter and Its Interactions	
Performance Standard	
PS1-MS-1 Develop models to describe the atomic composition of simple molecules and extended structures.	SE/TE: Components of Matter, 8-10 Molecules, 9 Hands-On Lab: Investigate, 9 Model It!: Molecules and Atoms, 9 Evidence-Based Assessment, 36-37 Model It!: Models of an Atom, 339 uEngineer It!: When Particles Collide, 367
Supporting Content	
PS1-MS-1.PS1.A Structure and Properties of Matter	
PS1-MS-1.PS1.A.i Substances are made from different types of atoms, which combine with one another in various ways. Atoms form molecules that range in size from two to thousands of atoms.	SE/TE: Elements, 8 Atoms, 8 Molecules, 9 Compounds, 10 Quest Connection, 334 The First Theories on Atoms, 336 Thomson's Model, 337 Rutherford's Model, 337-338 Bohr's Model, 338 Cloud Model, 339 Quest Connection, 358
PS1-MS-1.PS1.A.ii Solids may be formed from molecules, or they may be extended structures with repeating subunits (e.g., crystals).	SE/TE: Particles of a Solid, 48 Model It!: Crystalline and Amorphous Solids, 50 Types of Solids, 50

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PS1-MS-1.PS1.B Chemical Reactions	
PS1-MS-1.PS1.B.i Substances react chemically in characteristic ways. In a chemical process, the atoms that make up the original substances are regrouped into different molecules, and these new substances have different properties from those of the reactants.	SE/TE: Quest Kickoff: How can you use science to make special effects?, 2-3 Quest Connection, 24 Connect It!, 24 Examples of Chemical Change, 28 Quest Connection, 408 Chemical Change, 410 Building and Breaking Chemical Bonds, 411 Topic Review and Assess, 438-439
Performance Standard	
PS1-MS-2 Analyze and interpret data on the properties of substances before and after the substances interact to determine if a chemical reaction has occurred.	SE/TE: Quest Connection, 24 Chemical Changes in Matter, 27-29 Examples of Chemical Change, 28 Math Toolbox: Conservation of Mass, 29 Math Toolbox: Energy in Chemical Reactions, 31 Quest Connection, 368 Connect It!, 408 Chemical Change, 410 Energy Graphs for Chemical Reaction, 415 Other Factors, 417 Evidence-Based Assessment, 440-441 uDemonstrate Lab: Evidence of Chemical Change, 442-445

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Supporting Content	
PS1-MS-2.PS1.A Structure and Properties of Matter	
PS1-MS-2.PS1.A.i Each pure substance has characteristic physical and chemical properties (for any bulk quantity under given conditions) that can be used to identify it.	SE/TE: Quest Kickoff: How can you use science to make special effects?, 2-3 Quest Connection, 4 Connect It!, 4 Matter, 5-7 Physical Properties, 6 Chemical Properties, 7 Connect It!, 14 Quest Connection, 14 Weight, 15 Mass, 16 Topic Review and Assess, 34-35 Physical Properties of Solids, 49 Topic 8 Opener: Atoms and the Periodic Table, 330-331 Quest Kickoff: How can you use chemistry to solve a culinary mystery?, 332-333 Quest Connection, 346 Quest Check-In, 357 Lesson 2 Check, 418 Properties of Pure Substances, 431
Performance Standard	
PS1-MS-3 Gather and make sense of information to describe that synthetic materials come from natural resources and impact society.	SE/TE: Synthetic Materials, 429-432 Natural Resources as Building Blocks, 430 Properties of Pure Substances, 431 Accidental Synthetics Figure 2: Synthetic and How It Was Discovered, 431 The Impact of Synthetic Materials, 433-434 Synthetic Fibers, 433 Synthetic Foods, 433 Synthetic Fuels, 434 Synthetic Medicines, 434 Quest Check-In, 435 Topic Review and Assess, 438-439

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Supporting Content	
PS1-MS-3.PS1.A Structure and Properties of Matter	
PS1-MS-3.PS1.A.i Each pure substance has characteristic physical and chemical properties (for any bulk quantity under given conditions) that can be used to identify it.	SE/TE: Quest Kickoff: How can you use science to make special effects?, 2-3 Quest Connection, 4 Connect It!, 4 Matter, 5-7 Physical Properties, 6 Chemical Properties, 7 Connect It!, 14 Quest Connection, 14 Weight, 15 Mass, 16 Volume, 17 Examples of Chemical Change, 28 Physical Properties of Solids, 49 Topic 8 Opener: Atoms and the Periodic Table, 330-331 Quest Kickoff: How can you use chemistry to solve a culinary mystery?, 332-333 Quest Check-In, 357 Lesson 2 Check, 418 Properties of Pure Substances, 431
PS1-MS-3.PS1.B Chemical Reactions	
PS1-MS-3.PS1.B.i Substances react chemically in characteristic ways. In a chemical process, the atoms that make up the original substances are regrouped into different molecules, and these new substances have different properties from those of the reactants.	SE/TE: Quest Kickoff: How can you use science to make special effects?, 2-3 Quest Connection, 24 Connect It!, 24 Examples of Chemical Change, 28 Quest Connection, 408 Chemical Change, 410 Building and Breaking Chemical Bonds, 411 Topic Review and Assess, 438-439

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Performance Standard	
PS1-MS-4 Develop a model that predicts and describes changes in particle motion, temperature, and state of a pure substance when thermal energy is added or removed.	SE/TE: Changes of State, 25 Changes in Shape or Form, 26 Quest Kickoff: How can you use solids, liquids and gases to lift a car?, 44-45 Particles of a Solid, 48 Physical Properties of Liquids, 52 Physical Properties of Gases, 53 Particles of a Gas, 53 uEngineer It!: From "Ink" to Objects, 55 Quest Connection, 56 Connect It!, 56 Model It!: Dry Ice, 63 Model It!: Developing Models, 71 Topic Review and Assess, 78-79 Evidence-Based Assessment, 80-81 uDemonstrate Lab: Melting Ice, 82-85 Changing States, 143 Model It!: Develop Models, 144 Thermal Expansion, 161 Quest Check-In, 377 Topic Review and Assess, 438-439
Supporting Content	
PS1-MS-4.PS1.A Structure and Properties of Matter	
PS1-MS-4.PS1.A.i Gases and liquids are made of molecules or inert atoms that are moving about relative to each other.	SE/TE: Atoms, 8 Molecules, 9 Physical Properties of Liquids, 52 Physical Properties of Gases, 53 Particles of a Gas, 53

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<p>PS1-MS-4.PS1.A.ii In a liquid, the molecules are constantly in contact with others; in a gas, they are widely spaced except when they happen to collide. In a solid, atoms are closely spaced and may vibrate in position but do not change relative locations.</p>	<p>SE/TE: Atoms, 8 Molecules, 9 Hands-On Lab: Investigate, 47 Particles of a Solid, 48 A Ring of Solid Silver Figure 2, 48 Dancing in The Crowd Figure 3, 49 Particles of a Liquid, 51 Interactivity, 52 Particles of a Gas, 53 Interactivity, 53 Lesson 1 Check, 54 Quest Check-In, 54 Lesson 2 Check, 64 Topic Review and Assess, 78-79</p>
<p>PS1-MS-4.PS1.A.iii The changes of state that occur with variations in temperature or pressure can be described and predicted using these models of matter.</p>	<p>SE/TE: Quest Connection, 24 Connect It!, 24 Changes of State, 25 Changes in Shape or Form, 26 Interactivity, 58 Melting, 58 Changing Ice Into Water Figure 3, 58 Freezing, 59 Evaporation or Boiling? Figure 4, 60 The Effect of Pressure, 61 Pressure and Vaporization Figure 5 , 61 Video, 61 Condensation, 62 Hands-On Lab: Investigate, 62 Condensation on a Window Figure 6, 62 Changing State from Solid to Gas, 63 Model It!, 63 Lesson 2 Check, 64</p>

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PS1-MS-4.PS3.A Definitions of Energy	
PS1-MS-4.PS3.A.i The term “heat” as used in everyday language refers both to thermal energy (the motion of atoms or molecules within a substance) and the transfer of that thermal energy from one object to another. In science, heat is used only for this second meaning; it refers to the energy transferred due to the temperature difference between two objects.	<p>SE/TE: Temperature and Thermal Energy, 30 Thermal Energy, 57 Topic Review and Assess, 78-79 Energy in Motion and Force, 91-92 Nuclear Energy, 110 Thermal Energy, 111 Lesson 3 Check, 116 Quest Kickoff: How can you keep hot water from cooling down?, 138-139 Connect It!, 140 Quest Connection, 140 Thermal Energy and Heat, 141 Lesson 1 Check, 146 It's All Connected: Glassblowing, 147 Quest Connection, 148 Connect It!, 148 Types of Heat Transfer, 149-151 Thermal Energy and Work, 152-153 Energy Transformations, 152 Question It!, 153 Quest Check-In, 154 Lesson 2 Check, 154 uEngineer It!: Shockwave to the Future, 155 Quest Connection, 158 Connect It!, 158 Conductors and Insulators, 159 Thermal Properties of Materials, 159-161 Math Toolbox: Energy Change, Specific Heat, and Mass, 160 Thermal Expansion, 161 Lesson 3 Check, 165 Quest Check-In, 165 Topic Review and Assess, 166-167 uDemonstrate Lab: Testing Thermal Conductivity, 170-173</p>

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Performance Standard	
PS1-MS-5 Develop and use a model to describe how the total number of atoms does not change in a chemical reaction and thus mass is conserved.	SE/TE: Conservation of Mass, 28-29 Math Toolbox: Conservation of Mass, 29 Topic Review and Assess, 34-35 Topic Review and Assess, 166-167 Model It!, 422 Structure of an Equation, 422 Law of Conservation of Mass, 424-425 Lesson 3 Check, 427 Evidence-Based Assessment, 440-441
Supporting Content	
PS1-MS-5.PS1.B Chemical Reactions	
PS1-MS-5.PS1.B.i Substances react chemically in characteristic ways. In a chemical process, the atoms that make up the original substances are regrouped into different molecules, and these new substances have different properties from those of the reactants.	SE/TE: Quest Kickoff: How can you use science to make special effects?, 2-3 Quest Connection, 24 Connect It!, 24 Examples of Chemical Change, 28 Quest Connection, 408 Chemical Change, 410 Building and Breaking Chemical Bonds, 411 Topic Review and Assess, 438-439
PS1-MS-5.PS1.B.ii The total number of each type of atom is conserved, and thus the mass does not change.	SE/TE: Law of Conservation of Mass, 424-425 Interactivity, 424 Mass Conserved, 424 Video, 425 Interactivity, 425
Performance Standard	
PS1-MS-6 Undertake a design project to construct, test, and modify a device that either releases or absorbs thermal energy by chemical processes.	SE/TE: uDemonstrate Lab: Testing Thermal Conductivity, 170-173 Hands-On Lab: Demonstrate, 171

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Supporting Content	
PS1-MS-6.PS1.B Chemical Reactions	
PS1-MS-6.PS1.B.i Some chemical reactions release energy, others store energy.	SE/TE: Chemical Energy, 112 Lesson 3 Check, 116 Topic 9 Opener: Chemical Reactions, 394-395 Quest Kickoff: How can you design and build hot packs and cold packs?, 396-397 Quest Connection, 398 Exothermic Reaction, 414 Changes in Energy, 414 Endothermic Reaction, 414 Energy Graphs for Chemical Reaction, 415
PS1-MS-6.PS3.A Definitions of Energy	
PS1-MS-6.PS3.A.i The temperature of a system is proportional to the average internal kinetic energy and potential energy per atom or molecule (whichever is the appropriate building block for the system's material). The details of that relationship depend on the type of atom or molecule and the interactions among the atoms in the material. Temperature is not a direct measure of a system's total thermal energy. The total thermal energy (sometimes called total internal energy) of a system depends jointly on the temperature, the total number of atoms in the system, and the state of the material.	SE/TE: Temperature and Thermal Energy, 30 Thermal Energy, 57 Temperature, 57 Topic Review and Assess, 78-79 Temperature And Its Measurement, 142 How Temperature and Thermal Energy Are Related, 143-145 Comparing Thermal Energy, 144 Changes in Temperature, 145 Lesson 1 Check, 146 Question It!, 153 Quest Check-In, 154 Lesson 2 Check, 154 Lesson 3 Check, 165
PS1-MS-6.ETS1.B Developing Possible Solutions	
Performance Standard	
PS1-MS-6.ETS1.B.i A solution needs to be tested, and then modified on the basis of the test results in order to improve it.	SE/TE: Hands-On Lab: Redesign and Retest, 125 Go Online, 125 Test and Evaluate a Solution, 508 Redesign and Retest the Solution, 509

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Supporting Content	
PS1-MS-6.ETS1.B.ii The iterative process of testing the most promising solutions and modifying what is proposed on the basis of the test results leads to greater refinement and ultimately to an optimal solution.	SE/TE: uDemonstrate Lab: Making Waves, Analyze and Interpret Data, 3. Evaluate Your Tests, 235 Test and Evaluate a Solution, 508
PS2-MS Motion and Stability: Forces and Interactions	
Performance Expectation	
PS2-MS-1 Apply Newton’s Third Law to design a solution to a problem involving the motion of two colliding objects.	SE/TE: uEngineer It!: Generating Energy From Potholes, 479 Evidence-Based Assessment, 492-493 uDemonstrate Lab: Stopping on a Dime, 494-497
Supporting Content	
PS2-MS-1.PS2.A Forces and Motion	
PS2-MS-1.PS2.A.i For any pair of interacting objects, the force exerted by the first object on the second object is equal in strength to the force that the second object exerts on the first, but in the opposite direction (Newton’s third law).	SE/TE: Quest Kickoff: How can you build a complex machine to do something simple?, 88-89 Quest Check-In, 99 Quest Check-In, 106 Quest Connection, 108 Quest Check-In, 116 Quest Connection, 118 Quest Check-In, 125 Topic 10 Opener: Forces and Motion, 446-447 Quest Kickoff: How can you take the crash out of a collision?, 448-449 Types of Forces, 454 Lesson 1 Check, 457 Action-Reaction Pairs, 475 Balanced and Action-Reaction Forces, 476 Detecting Forces and Motion, 476 Newton’s Laws Together, 477 Lesson 3 Check, 478 Quest Check-In, 478 Connect It!, 480 Lesson 4 Check, 488 Topic Review and Assess, 490-491

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Performance Standard	
PS2-MS-2 Plan an investigation to provide evidence that the change in an object’s motion depends on the sum of the forces on the object and the mass of the object.	SE/TE: Math Toolbox: Effects of Net Force, 456 Connect It!, 470 Newton's First Law of Motion, 471 Newton's Second Law of Motion, 472-473 Balanced and Action-Reaction Forces, 476 Newton’s Laws Together, 477 Lesson 3 Check, 478 Quest Check-In, 478 Topic Review and Assess, 490-491 uDemonstrate Lab: Stopping on a Dime, 494-497
Support Content	
PS2-MS-2.PS2.A Forces and Motion	
PS2-MS-2.PS2.A.i The motion of an object is determined by the sum of the forces acting on it; if the total force on the object is not zero, its	SE/TE: Describing Force, 453 Balanced and Unbalanced Forces, 455-456

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motion will change. The greater the mass of the object, the greater the force needed to achieve the same change in motion. For any given object, a larger force causes a larger change in motion.	Quest Connection, 458 Newton's Second Law of Motion, 472-473 uDemonstrate Lab: Stopping on a Dime, 494-497
PS2-MS-2.PS2.A.ii All positions of objects and the directions of forces and motions must be described in an arbitrarily chosen reference frame and arbitrarily chosen units of size. In order to share information with other people, these choices must also be shared.	SE/TE: Reference Points, 451 Interactivity, 451 Movin' Along Figure 1, 451 Describing Force, 453 Representing Forces Figure 3, 453 Types of Forces, 454 Balanced and Unbalanced Forces, 455
Performance Standard	
PS2-MS-3 Ask questions about data to determine the factors that affect the strength of electric and magnetic forces.	SE/TE: Quest Connection, 240 Electric Force, Fields, and Energy, 241-243 Electric Force, 242 Lesson 1 Check, 248 uEngineer It!: Electromagnetism In Action, 265 Quest Connection, 266 Question It!: Types of Current, 271
Supporting Content	
PS2-MS-3.PS2.B Types of Interactions	
PS2-MS-3.PS2.B.i Electric and magnetic (electromagnetic) forces can be attractive or repulsive, and their sizes depend on the magnitudes of the charges, currents, or magnetic strengths involved and on the distances between the interacting objects.	SE/TE: uEngineer It!: Gathering Speed with Superconductors, 33 Quest Connection, 240 Electric Force, Fields, and Energy, 241-243 Electric Force, 242 Lesson 1 Check, 248 Quest Check-In, 248 Quest Connection, 250 Magnetic Force and Energy, 251-252 Magnetic Force, 252 Magnetic Fields, 253-256 Lesson 2 Check, 257 Quest Connection, 258 Electromagnetic Principles, 259 Magnetic Fields around Straight Wires, 260 Magnetic Fields and Current, 260-261

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Performance Standard	
PS2-MS-4 Construct and present arguments using evidence to support the claim that gravitational interactions are attractive and depend on the masses of interacting objects.	SE/TE: Universal Gravitation, 484 Universal Gravitation Figure 3, 484 Literacy Connection: Write Arguments, 485 Factors Affecting Gravity, 485 Weight and Mass, 485 Extraordinary Science: Spacetime Curvature and Gravitational Waves, 489 uDemonstrate Lab: Stopping on a Dime, 494-497
Supporting Content	
PS2-MS-4.PS2.B Types of Interactions	
PS2-MS-4.PS2.B.i Gravitational forces are always attractive. There is a gravitational force between any two masses, but it is very small except when one or both of the objects have	SE/TE: Types of Forces, 454 Universal Gravitation, 484 Video, 484

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large mass-e.g., Earth and the sun.	Lesson 4 Check, 488 Extraordinary Science: Spacetime Curvature and Gravitational Waves, 489
Performance Standard	
PS2-MS-5 Conduct an investigation and evaluate the experimental design to provide evidence that fields exist between objects exerting forces on each other even though the objects are not in contact.	SE/TE: Quest Connection, 240 Electric Force, Fields, and Energy, 241-243 Electric Force, 242 Electric Fields, 242 Lesson 1 Check, 248 Magnetic Force and Energy, 251-252 Magnetic Force, 252 Lesson 2 Check, 257 Magnetic Force on Moving Charges, 267-269 Loop of Current in a Magnetic Field, 268 How Generators Work, 273 Generators and Transformers, 273-274 How Transformers Work, 274 Topic Review and Assess, 278-279 uDemonstrate Lab: Planetary Detective, 282-285 Types of Forces, 454
Supporting Content	
PS2-MS-5.PS2.B Types of Interactions	
PS2-MS-5.PS2.B.i Forces that act at a distance (electric, magnetic, and gravitational) can be explained by fields that extend through space and can be mapped by their effect on a test object (a charged object, or a ball, respectively).	SE/TE: Electric Force, 242 Electric Fields, 242 Lesson 1 Check, 248 Quest Check-In, 248 Extraordinary Science: Bumblebees and Electric Flowers, 249 Magnetic Force and Energy, 251-252 Magnetic Force, 252 Lesson 2 Check, 257 Magnetic Force on Moving Charges, 267-269 Galvanometers, 268 Loop of Current in a Magnetic Field, 268 Electric Motors, 269 Electromagnetic Induction, 270-272 Induced Current and Moving Conductors, 270

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	Induced Current and Moving Magnets, 271 How Generators Work, 273 Generators and Transformers, 273-274 How Transformers Work, 274 Topic Review and Assess, 278-279
PS3-MS Energy	
Performance Standard	
PS3-MS-1 Construct and interpret graphical displays of data to describe the relationships of kinetic energy to the mass of an object and to the speed of an object.	SE/TE: Factors Affecting Kinetic Energy, 101 Kinetic Energy, 101-102 Calculating Kinetic Energy, 102 Math Toolbox: Mass, Speed, and Kinetic Energy, 102 Lesson 2 Check, 106 Lesson 4 Check, 125 Topic Review and Assess, 128-129
Supporting Content	
PS3-MS-1.PS3.A Definitions of Energy	
PS3-MS-1.PS3.A.i Motion energy is properly called kinetic energy; it is proportional to the mass of the moving object and grows with the square of its speed.	SE/TE: Factors Affecting Kinetic Energy, 101 Kinetic Energy, 101-102 Calculating Kinetic Energy, 102 Math Toolbox: Mass, Speed, and Kinetic Energy, 102 Lesson 2 Check, 106 Lesson 4 Check, 125 Topic Review and Assess, 128-129
Performance Standard	
PS3-MS-2 Develop a model to describe that when the arrangement of objects interacting at a distance changes, different amounts of potential energy are stored in the system.	SE/TE: Potential Energy, 103-105 Gravitational Potential Energy, 104 Elastic Potential Energy, 105 Lesson 2 Check, 106

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	Model It!: Conservation in Demolition, 121 Lesson 4 Check, 125 Topic Review and Assess, 128-129 Evidence-Based Assessment, 130-131 uDemonstrate Lab: 3, 2, 1...Liftoff!, 132-135 Question It!, 242 Charges and Potential Energy, 243 Lesson 1 Check, 248 uDemonstrate Lab: Planetary Detective, 282-285 Gravitational Potential Energy, 486 Model It!: Develop Models, 487 Forces and Motion, 487 Topic Review and Assess, 490-491 uDemonstrate Lab: Stopping on a Dime, 494-497
Supporting Content	
PS3-MS-2.PS3.A Definitions of Energy	
PS3-MS-2.PS3.A.i A system of objects may also contain stored (potential) energy, depending on their relative positions.	SE/TE: Potential Energy, 103-105 Gravitational Potential Energy, 104 Lesson 2 Check, 106 Model It!: Conservation in Demolition, 121 Lesson 4 Check, 125 Topic Review and Assess, 128-129 Question It!, 242 Gravitational Potential Energy, 486 Model It!: Develop Models, 487 Forces and Motion, 487 Topic Review and Assess, 490-491 uDemonstrate Lab: Stopping on a Dime, 494-497
PS3-MS-2.PS3.C Relationship Between Energy and Forces	

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PS3-MS-2.PS3.C.i When two objects interact, each one exerts a force on the other that can cause energy to be transferred to or from the object.	SE/TE: Energy Transformation and Transfer, 121 Reading Check: Cite Textual Evidence, 121 Lesson 4 Check, 125 Evidence-Based Assessment, 130-131 Analyze and Interpret Data, 135 Quest Connection, 458
Performance Standard	
PS3-MS-3 Apply scientific principles to design, construct, and test a device that either minimizes or maximizes thermal energy transfer.	SE/TE: uDemonstrate Lab: 3, 2, 1...Liftoff!, 132-135 Quest Kickoff: How can you keep hot water from cooling down?, 138-139 uEngineer It!: Shockwave to the Future, 155 uDemonstrate Lab: Testing Thermal Conductivity, 170-173 The Engineering and Design Process, 506-509
Supporting Content	
PS3-MS-3.PS3.A Definitions of Energy	
PS3-MS-3.PS3.A.i Temperature is a measure of the average kinetic energy of particles of matter. The relationship between the temperature and the total energy of a system depends on the types, states, and amounts of matter present.	SE/TE: Temperature, 57 Temperature and Its Measurement, 142 How Temperature and Thermal Energy Are Related, 143-145 Changing States, 143 Comparing Thermal Energy, 144 Changes in Temperature, 145
PS3-MS-3.PS3.B Conservation of Energy and Energy Transfer	
PS3-MS-3.PS3.B.i Energy is spontaneously transferred out of hotter regions or objects and into colder ones.	SE/TE: Types of Heat Transfer , 149 Heat Flow Figure 2, 150-151 uDemonstrate Lab: Testing Thermal Conductivity, 170-173

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PS3-MS-3.ETS1.A Defining and Delimiting an Engineering Problem	
PS3-MS-3.ETS1.A.i The more precisely a design task's criteria and constraints can be defined, the more likely it is that the designed solution will be successful. Specification of constraints includes consideration of scientific principles and other relevant knowledge that is likely to limit possible solutions.	SE/TE: uEngineer It! Defining the Problem: STEM, making Water Safe to Drink, Design Challenge, 407 uEngineer It! Defining the Problem: STEM, Generating Energy from Potholes, Design Challenge, 479 The Engineering and Design Process, 506 Define the Problem, 506-507 Reflect, 506 Make Meaning, 507 Make Meaning, 508
PS3-MS-3.ETS1.B Developing Possible Solutions	SE/TE: uEngineer It! Designing Solutions: STEM, From "Ink" to Objects: 3D Printing, Design Challenge, 55 Develop Possible Solutions, 507 Design a Solution, 508 Test and Evaluate a Solution, 508 Communicate the Solution, 509 Redesign and Retest the Solution, 509
PS3-MS-3.ETS1.B.i A solution needs to be tested, and then modified on the basis of the test results in order to improve it. There are systematic processes for evaluating solutions with respect to how well they meet criteria and constraints of a problem.	SE/TE: Test and Evaluate a Solution, 508 Redesign and Retest the Solution, 509 Communicate , 509
Performance Standard	
PS3-MS-4 Plan an investigation to determine the relationships among the energy transferred, the type of matter, the mass, and the change in the average kinetic energy of the particles as measured by the temperature of the sample.	SE/TE: Temperature and Thermal Energy, 30 Thermal Energy and Changes in Matter, 31 Temperature, 57 How Temperature and Thermal Energy Are Related, 143-145 Comparing Thermal Energy, 144 Changes in Temperature, 145 Literacy Connection: Conduct Research Projects, 153

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	uDemonstrate Lab: Testing Thermal Conductivity, 170-173
Supporting Content	
PS3-MS-4.PS3.A Definitions of Energy	
PS3-MS-4.PS3.A.i Temperature is a measure of the average kinetic energy of particles of matter. The relationship between the temperature and the total energy of a system depends on the types, states, and amounts of matter present.	SE/TE: Temperature, 57 Temperature And Its Measurement, 142 How Temperature and Thermal Energy Are Related, 143-145 Changing States, 143 Comparing Thermal Energy, 144 Changes in Temperature, 145
PS3-MS-4.PS3.B Conservation of Energy and Energy Transfer	
PS3-MS-4.PS3.B.i The amount of energy transfer needed to change the temperature of a matter sample by a given amount depends on the nature of the matter, the size of the sample, and the environment.	SE/TE: Thermal Energy and Changes in Matter, 31 How Temperature and Thermal Energy Are Related, 143-145 Comparing Thermal Energy, 144 Changes in Temperature, 145 uDemonstrate Lab: Testing Thermal Conductivity, 170-173
Performance Standard	
PS3-MS-5 Construct, use, and present arguments to support the claim that when the kinetic energy of an object changes, energy is transferred to or from the object.	SE/TE: Literacy Connection: Cite Textual Evidence, 110 Energy Changes Form, 119-121 Kinetic and Potential Energy, 120 Energy Transformation and Transfer, 121 Lesson 4 Check, 125 Topic Review and Assess, 128-129 Evidence-Based Assessment, 130-131 uDemonstrate Lab: 3, 2, 1...Liftoff!, 132-135 uEngineer It!: Shockwave to the Future, 155 Plan It!: Materials for Airplanes, 164 Evidence-Based Assessment, 168-169
Supporting Content	
PS3-MS-5.PS3.B Conservation of Energy and Energy Transfer	
PS3-MS-5.PS3.B.i When the motion energy of an object changes, there is inevitably some	SE/TE: Motion and Energy, 91

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other change in energy at the same time.	Energy Changes Form, 119-121 Kinetic and Potential Energy, 120 Conservation of Energy in Transfers, 123 Topic Review and Assess, 128-129
PS4-MS Waves	
Performance Standard	
PS4-MS-1 Use mathematical representations to describe a simple model for waves that includes how the amplitude of a wave is related to the energy in a wave.	SE/TE: Types of Waves, 179-181 Properties of Waves, 182-183 Properties of Waves Figure 5, 182 Math Toolbox: Wave Properties, 184 Wave Energy, 184 Lesson 1 Check, 185 Connect It!, 198
PS4-MS-1.PS4.A Wave Properties	
PS4-MS-1.PS4.A.i A simple wave has a repeating pattern with a specific wavelength, frequency, and amplitude.	SE/TE: Types of Waves, 179-181 Interactivity, 181 Properties of Waves, 182-183 Properties of Waves Figure 5, 182 Lesson 1 Check, 185 Math Toolbox: Frequencies and Wavelengths of Light, 213
Performance Standard	
PS4-MS-2 Develop and use a model to describe that waves are reflected, absorbed, or transmitted through various materials.	SE/TE: Topic 5 Opener: Waves and Electromagnetic Radiation, 174-175 Quest Connection, 178 Reflection, Refraction and Absorption, 189-191 Plan It!: Develop Models, 190 uEngineer It!: Say "Cheese!", 197

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	Connect It!, 198 The Behavior of Sound, 199-201 Absorption, 200 Reflection and Transmission, 200 Diffraction, 201 Model It!, 201 Lesson 3 Check, 207 Quest Connection, 208 Model It!: Polarizing Glasses, 211 Quest Check-In, 216 Model It!: Fun with Mirrors, 224 Quest Check-In, 227 Topic Review and Assess, 228-229 Evidence-Based Assessment, 230-231 uDemonstrate Lab: Making Waves, 232-235
Supporting Content	
PS4-MS-2.PS4.A Wave Properties	
PS4-MS-2.PS4.A.i A sound wave needs a medium through which it is transmitted.	SE/TE: Case Study: Sound and Light at the Ballpark, 186-187 The Behavior of Sound, 199-201 Diffraction, 201 Model It!, 201 Density, 202 Stiffness, 202 Lesson 3 Check, 207
PS4-MS-2.PS4.B Electromagnetic Radiation	
PS4-MS-2.PS4.B.i When light shines on an object, it is reflected, absorbed, or transmitted through the object, depending on the object's material and the frequency (color) of the light.	SE/TE: Quest Connection, 188 Reflection, Refraction and Absorption, 189-191 Question It!: Classify, 191

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	Quest Check-In, 196 Connect It!, 208 Models of Electromagnetic Wave Behavior, 210-211 Quest Check-In, 216 Quest Connection, 218 Color Filters, 221 Reflecting Light, 222-224 Mirror Images, 223 Concave Mirrors, 224
PS4-MS-2.PS4.B.ii The path that light travels can be traced as straight lines, except at surfaces between different transparent materials (e.g., air and water, air and glass) where the light path bends.	SE/TE: Quest Connection, 178 Quest Check-In, 185 Connect It!, 188 Reflection, Refraction and Absorption, 189-191 Question It!: Classify, 191 Lesson 2 Check, 196 Quest Check-In, 196
PS4-MS-2.PS4.B.iii A wave model of light is useful for explaining brightness, color, and the frequency-dependent bending of light at a surface between media.	SE/TE: Properties of Waves, 182-183 Math Toolbox: Frequencies and Wavelengths of Light, 213 Lesson 3 Check, 216 Careers: Lighting Designer, 217 Color Filters, 221
PS4-MS-2.PS4.B.iv However, because light can travel through space, it cannot be a matter wave, like sound or water waves.	SE/TE: Quest Connection, 178 Types of Waves, 179 Case Study: Sound and Light at the Ballpark, 186-187 Characteristics of Electromagnetic Waves, 209

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Performance Standard	
<p>PS4-MS-3 Integrate qualitative scientific and technical information to support the claim that digitized signals are a more reliable way to encode and transmit information than analog signals.</p>	<p>SE/TE: uEngineer It!: A Life-Saving Mistake, 299 Quest Connection, 300 Signals and Information, 301-303 Electronic Signals, 302 Electromagnetic Signals, 303 Analog and Digital Signals, 304-306 Analog Signals, 304 Digital Signals, 304 Sampling Rate, 305 Binary Signals, 306 Transmitting Signals, 307-308 Sound Information, 307 Visual Information, 308 Lesson 2 Check, 309 Quest Check-In, 309 Case Study: Super Ultra High Definition!, 310-</p>

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Supporting Content	
PS4-MS-3.PS4.C Information Technologies and Instrumentation	
PS4-MS-3.PS4.C.i Digitized signals (sent as wave pulses) are a more reliable way to encode and transmit information.	SE/TE: Signals and Information, 301-303 Electronic Signals, 302 Electromagnetic Signals, 303 Analog and Digital Signals, 304-306 Analog Signals, 304 Digital Signals, 304 Sampling Rate, 305 Binary Signals, 306 Transmitting Signals, 307-308 Sound Information, 307 Lesson 2 Check, 309